

Additions and Corrections

Site-Specific Incorporation of the Alkaline Labile, Oxidative Stress Product (5R)-5,6-Dihydro-5-hydroxythymidine in an Oligonucleotide [*J. Am. Chem. Soc.*]. TRACY J. MATRAY AND MARC M. GREENBERG*

Page 6932: The following list of reagents belongs with Scheme 1.

(a) OsO₄, *N*-methylmorpholine *N*-oxide, *t*-BuOH, H₂O, 45 °C. (b) *m*-(Trifluoromethyl)benzoyl chloride, DMAP, THF, -10 °C. (c) *N*-methylcarbazole, THF, H₂O (9:1), *hν*. (d) *t*-BuMgCl (3.0 equiv), THF, 25 °C then allylpyrocarbonate (5.0 equiv). (e) AcOH (0.5 M), TBAF (0.5 M), THF, 25 °C. (f) (Allyloxy)bis(diisopropylamino)phosphine (1.2 equiv), diisopropylammonium tetrazolide (1.3 equiv), CH₂Cl₂, 25 °C.

Deferred Carbonylative Esterification in the Pd-Catalyzed Cyclic Carbometalation–Carbonylation Cascade [*J. Am. Chem. Soc.* 1994, 116, 7923]. TAKUMICHI SUGIHARA, CHRISTOPHE COPÉRET, ZBYSLAW OWCZARZYK, LORI S. HARRING, AND EI-ICHI NEGISHI*

Page 7924: The first sentence of the third paragraph should read as follows: Fourth, since the reaction of PhI with even 1 atm of CO in the presence of 5 mol % of Cl₂Pd(PPh₃)₂, 4-octyne (1 equiv), and MeOH (4 equiv) merely gives methyl benzoate in 73% yield without a sign of carbopalladation or acylpalladation, CO insertion is considerably faster than intermolecular carbopalladation.

Planar Tetracoordinate Carbons in Dimetallic Complexes: Quantum Chemical Investigations [*J. Am. Chem. Soc.* 1994, 116, 8259–8265]. CHARLES N. POUMBGA, MARC BÉNARD,* AND ISABELLA HYLAKRYSPIN

It has been stated that “no quantum chemical analysis of the electronic structure of [V₂(DMP)₄] has been carried out to date”. This was not exact, since a Fenske–Hall calculation reporting the σ²π²δ² nature of the V–V triple bond had been published previously. The reference is as follows: Cotton, F. A.; Diebold, M. P.; Shim, I. *Inorg. Chem.* 1985, 24, 1510–1516.

Highly Enantioselective, Low-Temperature Epoxidation of Styrene [*J. Am. Chem. Soc.* 1994, 116, 9333]. MICHAEL PALUCKI, PAUL J. POSPISIL, WEI ZHANG, AND ERIC N. JACOBSEN*

Equation 2: The equation should read

$$ee_{\text{obs}} = (ee_{\text{cis}} \times \% \text{ cis}) - (ee_{\text{trans}} \times \% \text{ trans})$$

The correct equation was used in all of the calculations described in the text and in Table 2, so this change does not affect any of the conclusions of the paper.

“Redox Tautomerism” in High-Valent Metal–oxo–aquo Complexes. Origin of the Oxygen Atom in Epoxidation Reactions Catalyzed by Water-Soluble Metalloporphyrins [*J. Am. Chem. Soc.* 1994, 116, 9375–9376]. JEAN BERNADOU,* ANNE-SYLVIE FABIANO, ANNE ROBERT, AND BERNARD MEUNIER*

The legend for Figure 1 should read as follows: Correlation between the content of H₂¹⁸O (*X*-axis) in the reaction mixture and the amount of labeled oxygen found in CBZ oxide (*Y*-axis) (see ref 10 for experimental conditions, MnTMPyP being the catalyst).

A Hypothetical Dense 3,4-Connected Carbon Net and Related B₂C and CN₂ Nets Built from 1,4-Cyclohexadienoid Units [*J. Am. Chem. Soc.* 1994, 116, 11456–11464]. MICHAEL J. BUCKNUM AND ROALD HOFFMANN*

The following supplementary material paragraph should have appeared with this article. This supplementary material appears with the microfiche and microfilm material for the December 14 issue (No. 25).

Supplementary Material Available: Tables of theoretical diffraction patterns and calculated Bragg angles (3 pages). This material is contained in many libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.